



XANES analysis and high energy resolution techniques

Jorge Moncada Beamline Scientist – ISS Beamline

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Agenda

- Fundamentals of XANES
- Relevance of the technique
- Interpretation of XANES
- High energy resolution techniques
- Data analysis

Acknowledgment

The following talk is heavily inspired by Denis Leshchev talks on XANES analysis and High-Resolution Spectroscopy talks at the 2024 NSLS-II EXAFS Workshop.

EXAFS in WoS 21,823 indexed entries and counting...



Source: Clarivate Web of Science. March 2025.

XANES in WoS 16,981 indexed entries and counting...



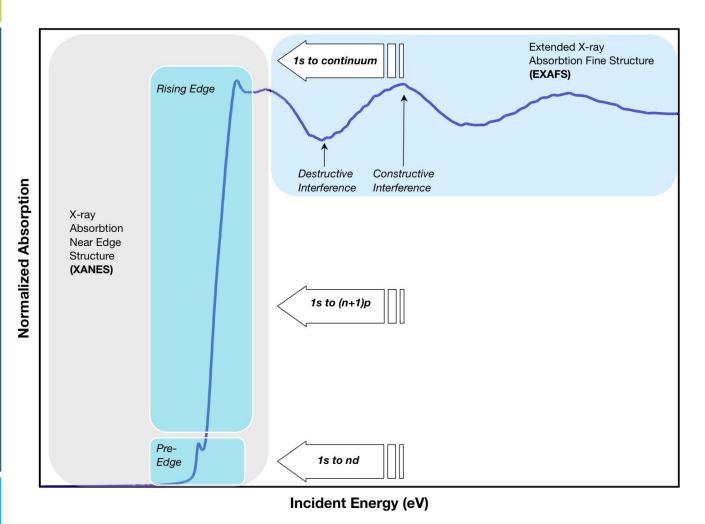
Fundamentals

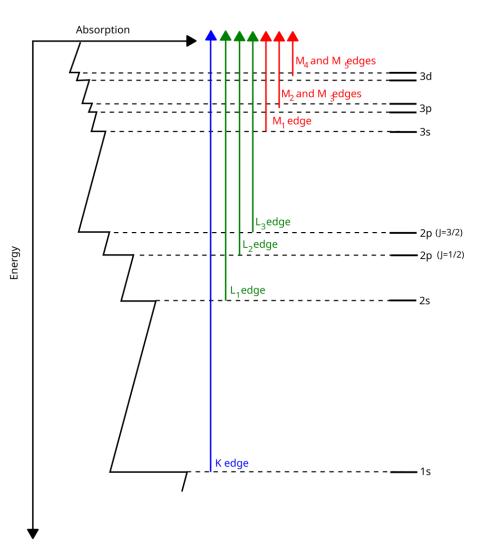


Relevance of XANES

- On the pre-edge region:
 - Ligand field strength and geometry
 - Spin state
 - Centrosymmetry
 - Metal-ligand overlap and covalency
- On the rising-edge region:
 - Geometric structure
 - Metal-ligand overlap via shakedown transitions
 - Charge on the metal center
- Fast data acquisition, high signal to noise ratio, data integrity at high temperature.

Fundamentals The XAFS Spectrum



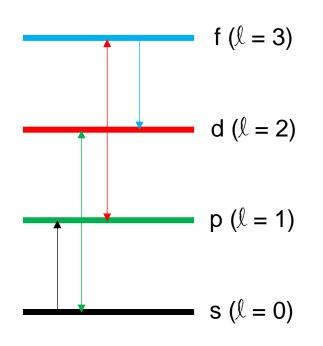


By Atenderholt and Muzarin at English Wikipedia, CC BY-SA 3.0.

Fundamentals

Nomenclature

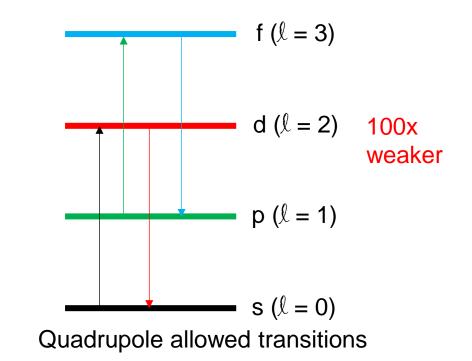
- The XANES spectrum is governed by the dipole selection rule:
 - A photon has an angular momentum (ℓ) of 1, so the electron on the absorbing atom must undergo $\Delta \ell = \pm 1$ to conserve momentum.
 - Dipole forbidden but quadrupole allowed transitions ($\Delta \ell = \pm 2$) may be present but much weaker



Dipole allowed transitions

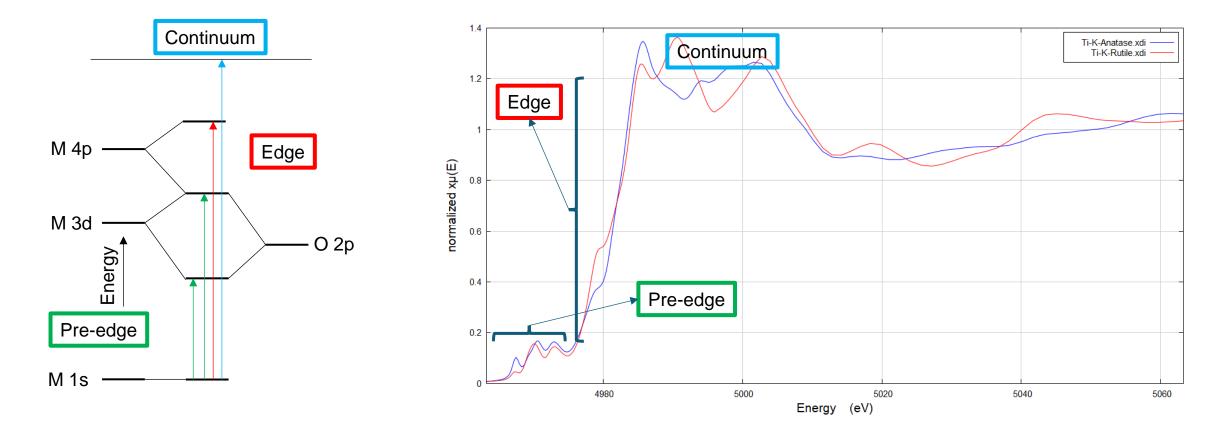
Dipole allowed transitions:

- s→p
 p→s, d
- d→p, f
- For K-edge: 1s to p (unoccupied) For L1-edge: 2s to p (unoccupied) For L2-edge: $2p_{1/2}$ to d (unoccupied) L3-edge: $2p_{3/2}$ to d (unoccupied)



Fundamentals

Metal K-edge XANES

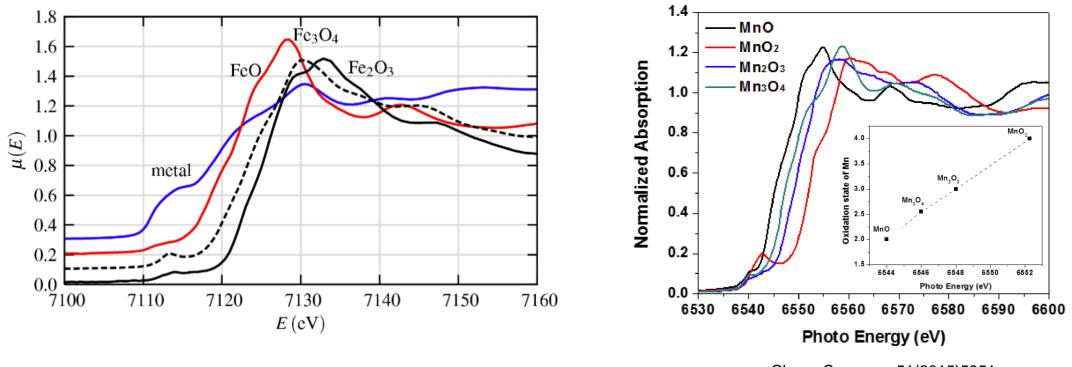


Metal K-pre-edge absorptions arise due to quadrupole-allowed dipole-forbidden 1s to 3d excitation ($\Delta \ell = \pm 2$, week) Metal K-rising-edge absorptions are dielectric dipole allowed 1s to 4p excitation ($\Delta \ell = \pm 1$, strong)

Sensitivity to electronic structure



Selectivity to oxidation state

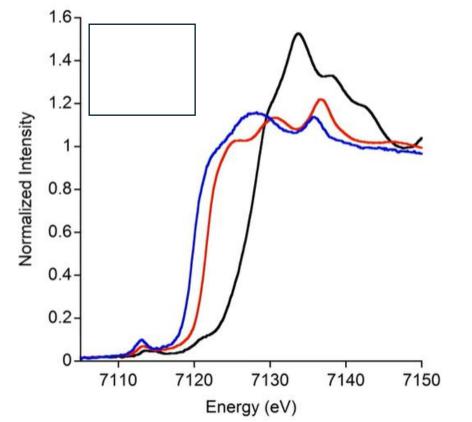


Fundamentals of XAFS, Matt Newville

Chem. Commun. 51(2015)5951

Rising edge and white line maxima shift to higher energy with increase of oxidation state*
 *Special considerations apply

Selectivity to bonding



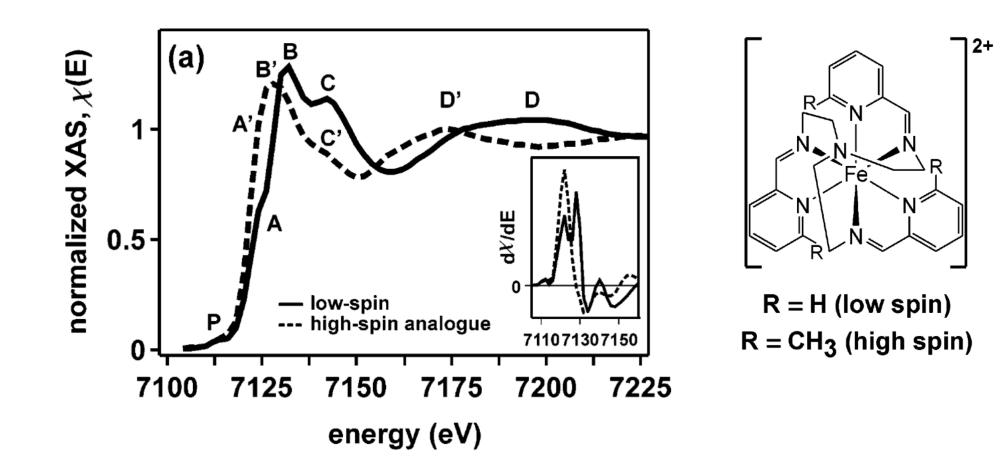
Chris Pollock, Fundamentals of X-ray Absorption Spectroscopy

Electronegativity of bounded halogens change the core energy level of Fe 1s electrons

Increase in electronegativity pulls electrons from Fe making look more oxidized

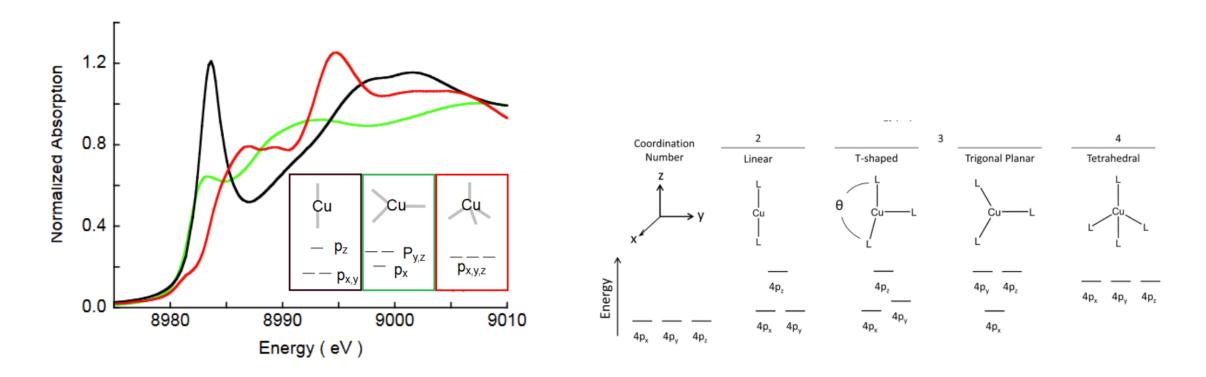
In general, heavier atoms push the edge lower in energy

Local structure



• Same oxidation state (Fe²⁺) but different spin shift

Local symmetry



M.L. Baker et al. / Coordination Chemistry Reviews 345 (2017) 182–208

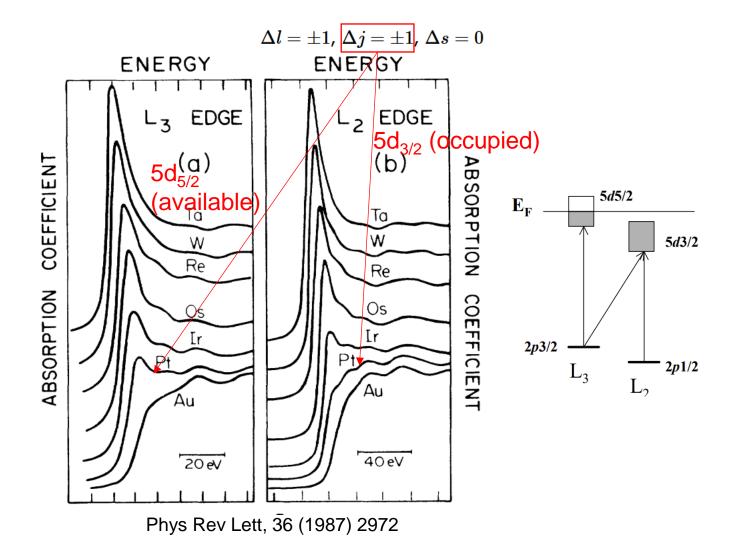
 $1s \rightarrow 4p$ transition and its energy and intensity depend on the ligand environment around the Cu(I) centers

L2 and L3 edges for 5d metals

For L2 and L3 edges - Number of holes in the 5d orbital (2p – 5d) determines the white line intensity

White line absent in gold because of full 5d shell

Absence of white line in Au-L3 and Pt-L2 edges frequently used to determine formation of intermetallic Au-X and Pt-X systems by charge transfer contribution to white line

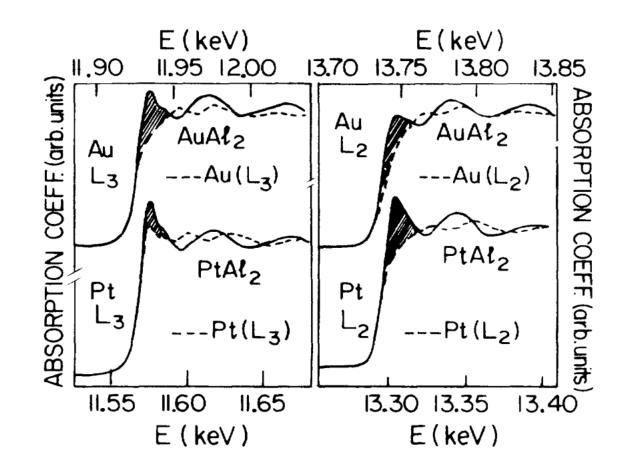


L2 and L3 edges for 5d metals

For L2 and L3 edges - Number of holes in the 5d orbital (2p – 5d) determines the white line intensity

White line absent in gold because of full 5d shell in L3-edge

Absence of white line in Au-L3 and Pt-L2 edges to determine formation of intermetallic AuAl₂ and PtAl₂ systems by charge transfer contribution to white line

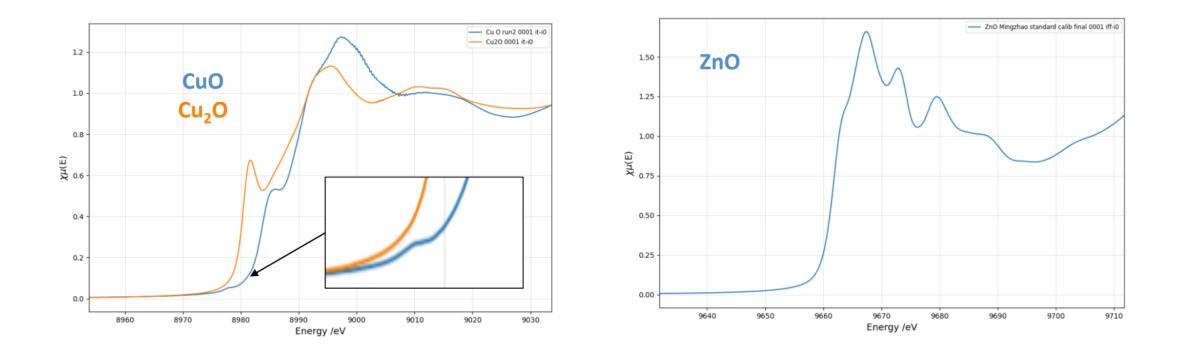


Phys Rev Lett, 36 (1987) 2972

Pre-edge



Pre-edge features

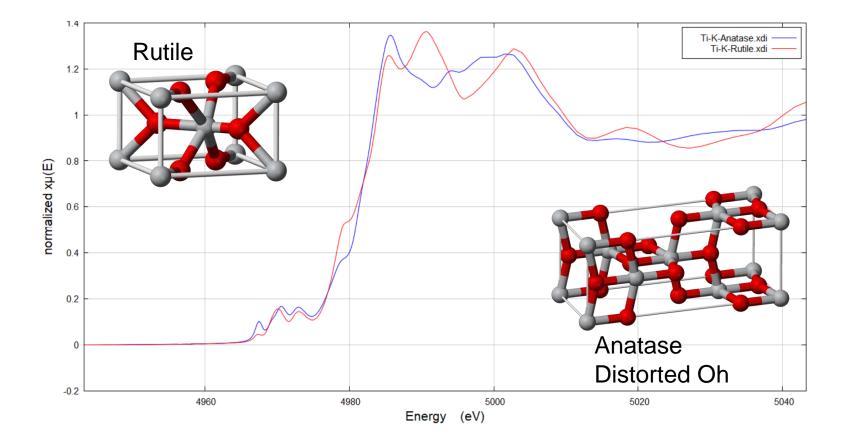


 Cu_2O , ZnO - d¹⁰ systems - do not have any pre-edge CuO - d⁹ system - has one!

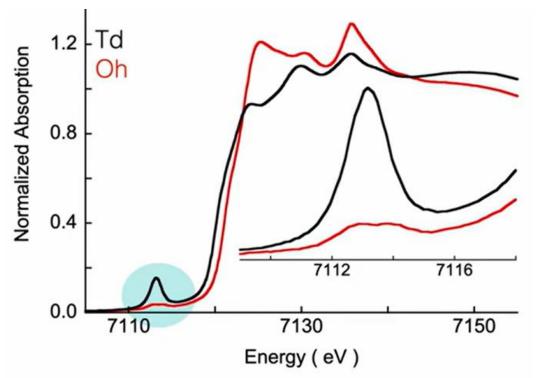
Pre-edge features

Rutile and anatase: Both TiO2 with Oh coordination and same oxidation state

Deviation from centrosymmetric geometry increase intensity of pre-edge features

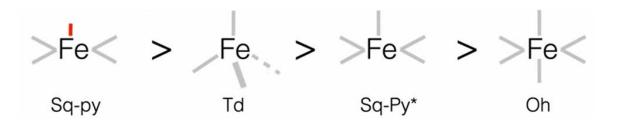


Pre-edge features



Ritimukta Sarangi, Electronic Structure Interpretation from XANES

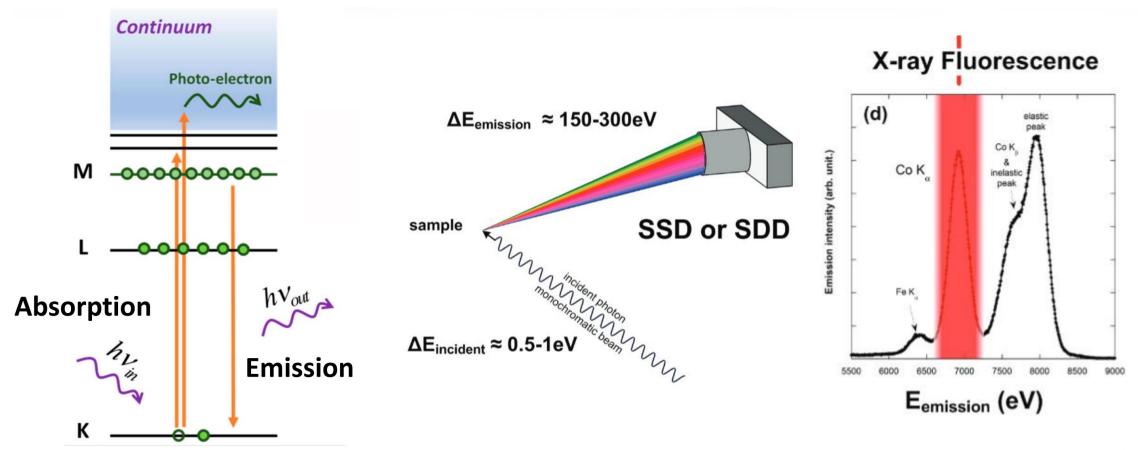
 Pre-edge intensity proportional to deviation from centrosymmetry (metal 3d-4p mixing)



A brief introduction

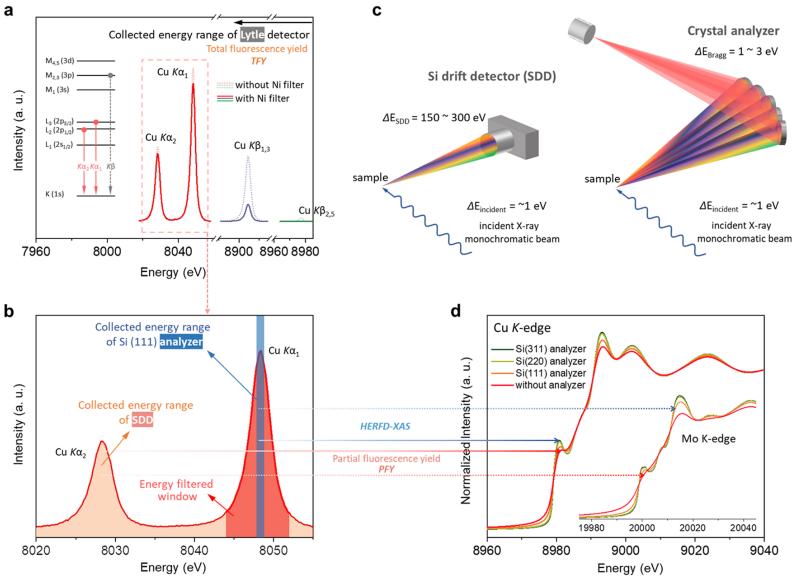


XAS with fluorescence detection



Biochimica et Biophysica Acta 1853 (2015) 1406–1415

J. Environ. Qual. 46:1146–1157 (2017).



Detection with narrow bandwidth (lower than core hole lifetime broadening).

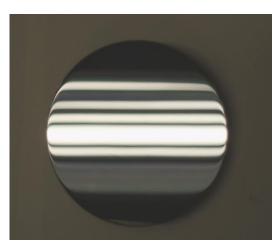
Fluorescence photons emitted from the sample are dispersed using curved Si/Ge crystals

Johann geometry

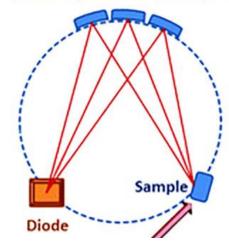
- Focusing geometry
- Spherical crystal curvature
- Dilute samples

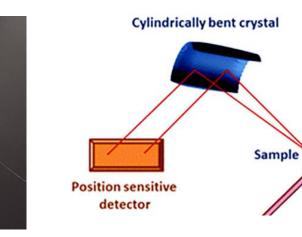
von Hamos geometry

- Dispersive geometry
- Full emission line in 1 image
- Cylindrical curvature
- Concentrated samples
- Fast

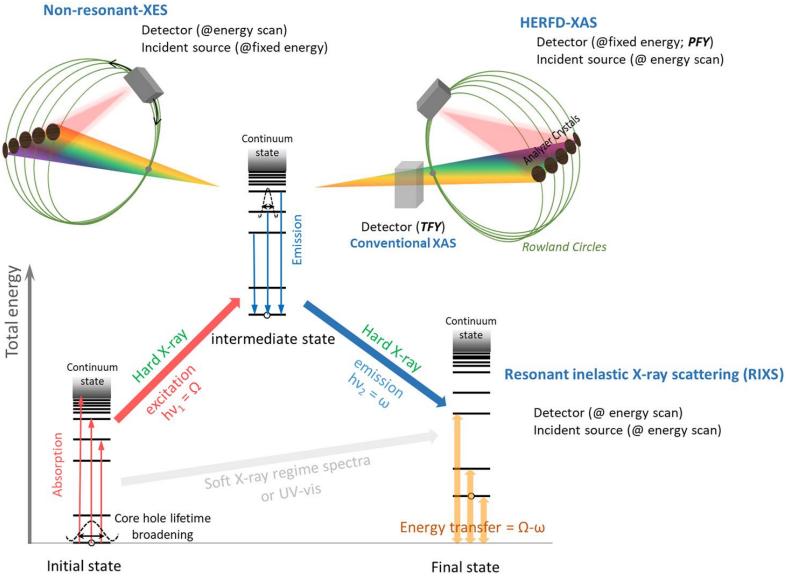


Spherically bent analyzer crystals

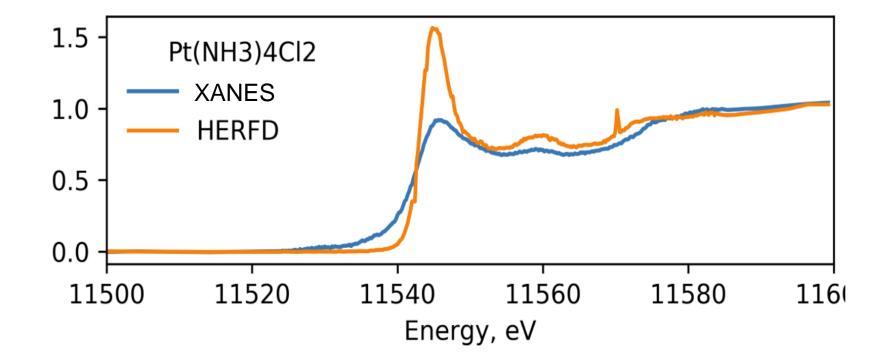




Total energy schemes for various X-ray photon-in (Ω) and photon-out (ω) spectroscopies, accompanied with their brief fundamentals and experimental schemes.

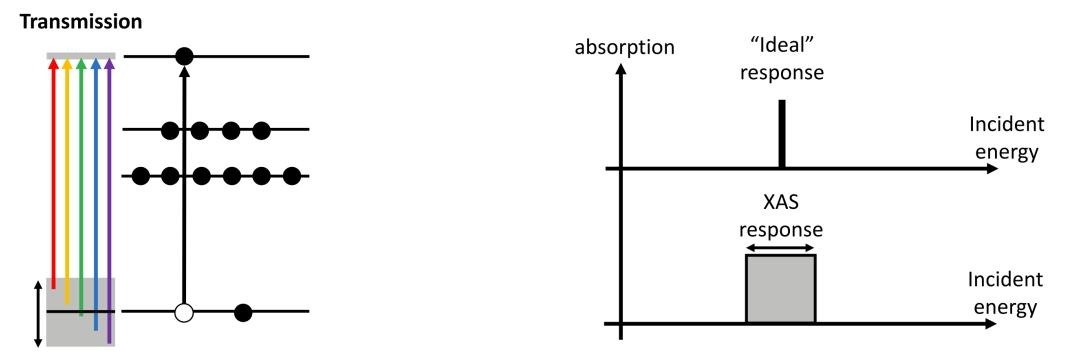


High-energy resolution fluorescence detected (HERFD) XAS



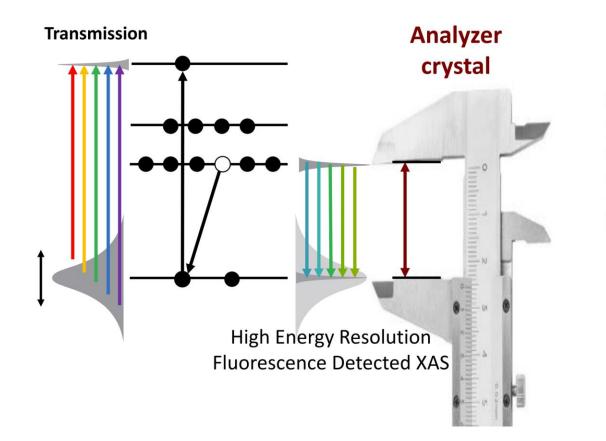
Helps to overcome core hole broadening!

High-energy resolution fluorescence detected (HERFD) XAS

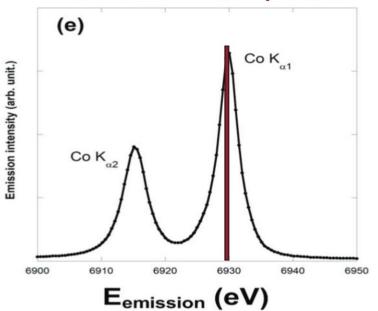


Finite core hole lifetime results in the energy broadening of the level

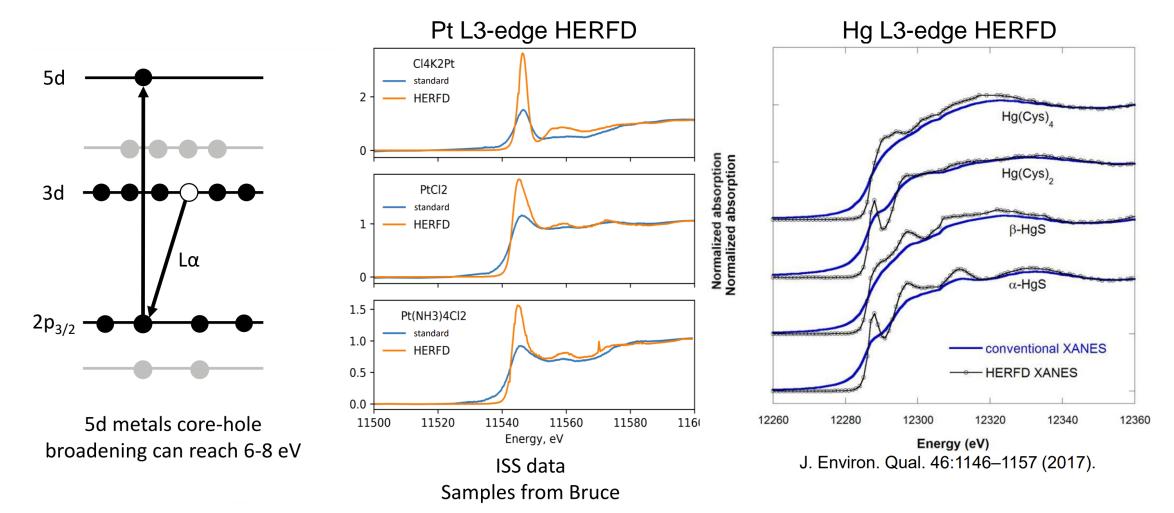
High-energy resolution fluorescence detected (HERFD) XAS



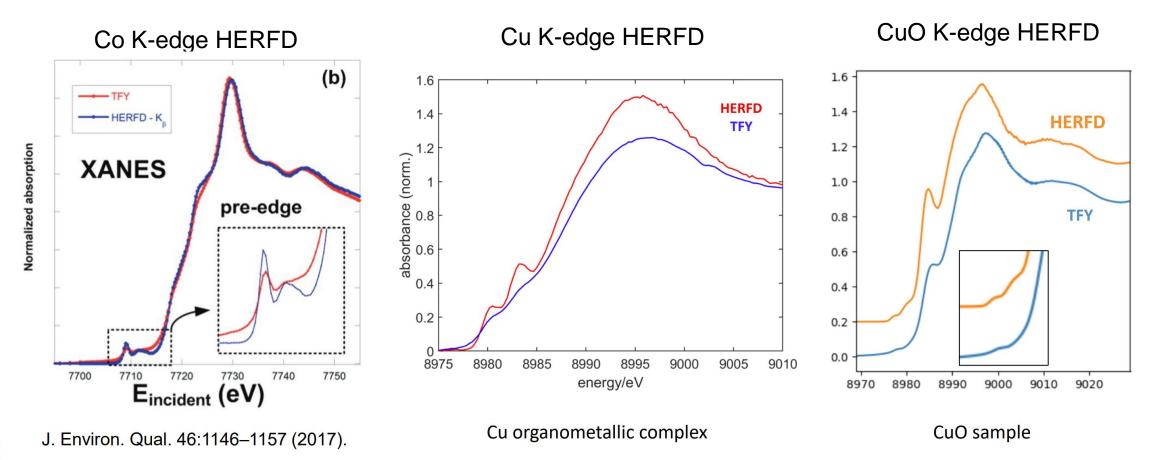




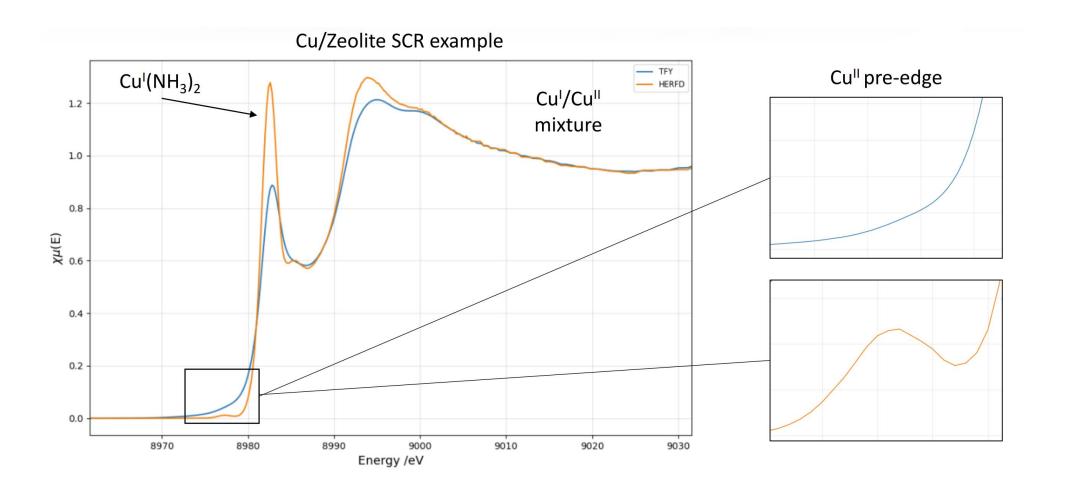
HERFD Examples: 5d metals



HERFD Examples: 3d metals

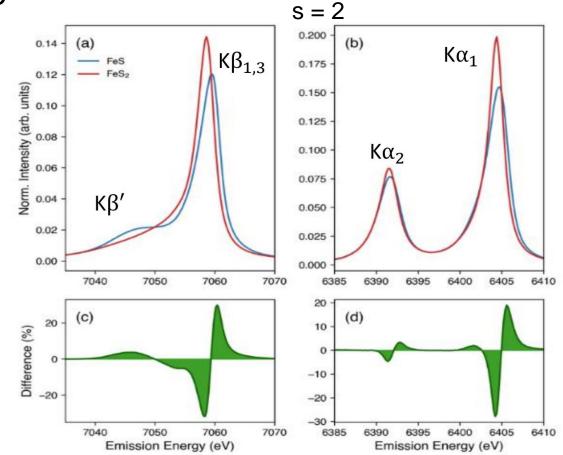


HERFD Examples: 3d metals



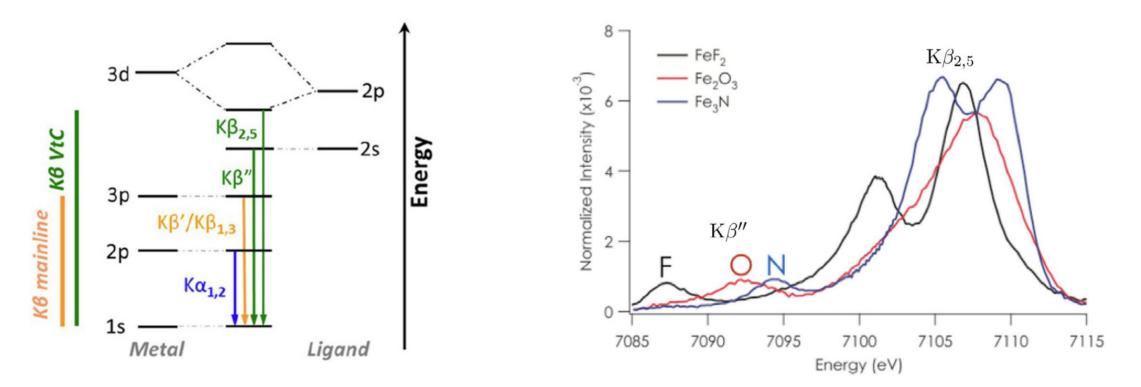
X-ray emission spectroscopy: a spin probe

- Fix $\rm E_{in}$ well above the edge (100-150 eV), scan $\rm E_{out}$
- Kα (2p → 1s) and Kβ (3p → 1s) lines are sensitive to spin state of the absorbing atom (3d/2p and 3d/3p exchange interaction)



Inorg. Chem.2020, 59, 12518-12535

X-ray emission spectroscopy: valence-to-core



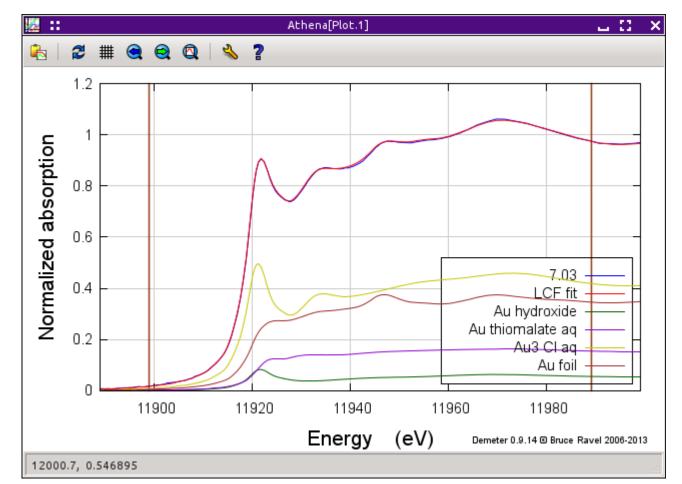
Coordination Chemistry Reviews, 423(2020)213466



- Pre-edge and raising edge sensitive to:
 - Oxidation state
 - Spin
 - Multiplet structure
 - Symmetry
 - Bond length
 - Covalency
- Software for data analysis:
 - Demeter Athena
 - Larch Larix
- Software for XANES calculations:
 - FEFF
 - Ocean
 - MXAN
 - DFT

Analysis of mixtures – Linear Combination Fitting

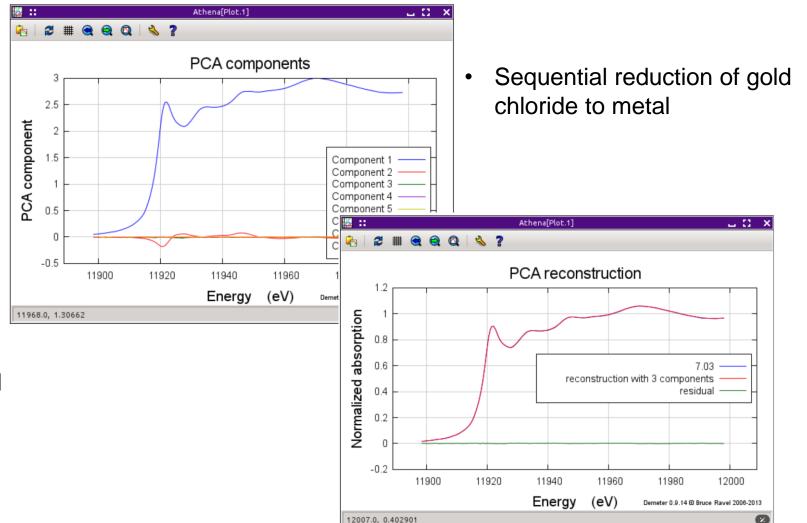
- Real systems often contain more than one specie
- Composition of mixtures can be analyzed by using reference spectra
- As total contributions are the sum of individual ones, LCF can be used to quantify the extent of each contribution
- If many components are present in the sample, you might need to run many different combinations to figure out which one fits the best



Bruce Ravel, Athena documentation

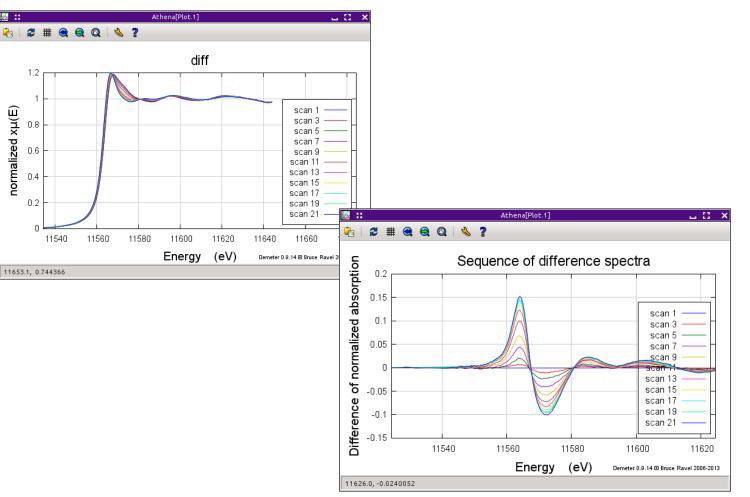
Analysis of mixtures – Principal Components Analysis

- Decompose a series of spectra into an orthogonal set of eigenvectors
- Number of eigenvalues equals the number of physical components of the system
- Target transform can be used to test potential physical components against the eigenvectors
- If dataset is not properly aligned and normalized, additional eigenvectors non-eligible eigenvectors may show up



Analysis of mixtures – Difference Spectra

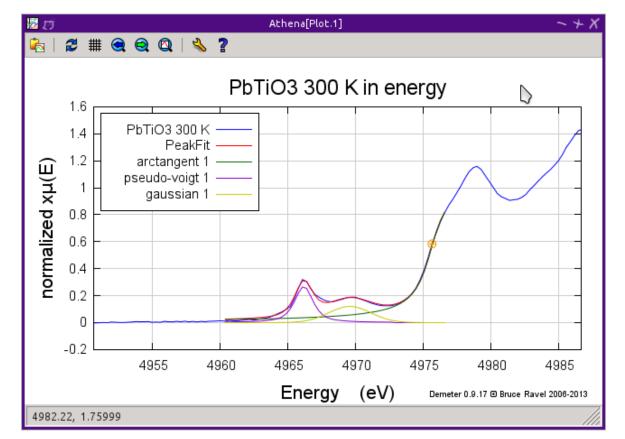
- Subtraction of a series of spectra by initial (reference) spectrum
- Commonly used to highlight subtle changes in a data sequence
- Provided example for Pt sequential coverage by hydrogen



Bruce Ravel, Athena documentation

Data analysis Analysis of mixtures – Peak fitting

- Spectrum deconstruction into a set of step-like and peak functions
- More meaningful when performed across related data
- Main drawback: physical significance of the line-shapes is not well understood



Bruce Ravel, Athena documentation

Final remarks

XANES – what can we get?

- Oxidation state
- Spin
- Multiplet structure
- Symmetry
- Covalency
- Bond lengths



EXAFS

- Larger signal can be collected at lower concentration
- Easier to crudely interpret fingerprinting
- Harder to fully interpret big efford needed for modeling/fitting spectra features

On high energy resolution techniques

- Techniques as HERFD-XAS, XES, and RIXS result complimentary on providing insights into electronic structure of materials
- HERFD-XAS: cleaner data, enhanced sensitivity
- XES: spin, ligand speciation
- RIXS: covalency, bonding
- Special care on avoiding sample damage

More questions?



